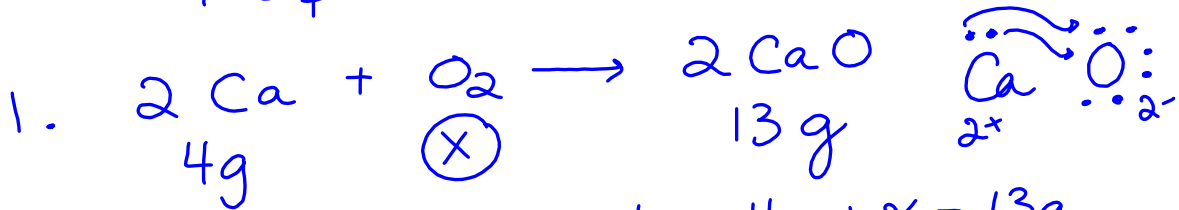


For each of the following reactions :

1) Find the missing mass.

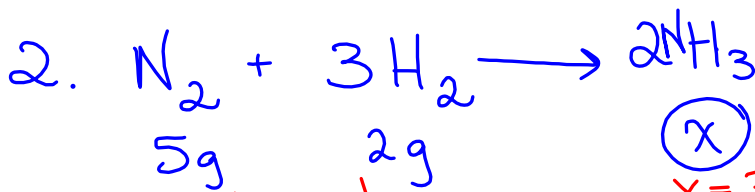
2) Draw a Lewis Dot notation for the products.



Total mass of reactants = 4g + x = 13g

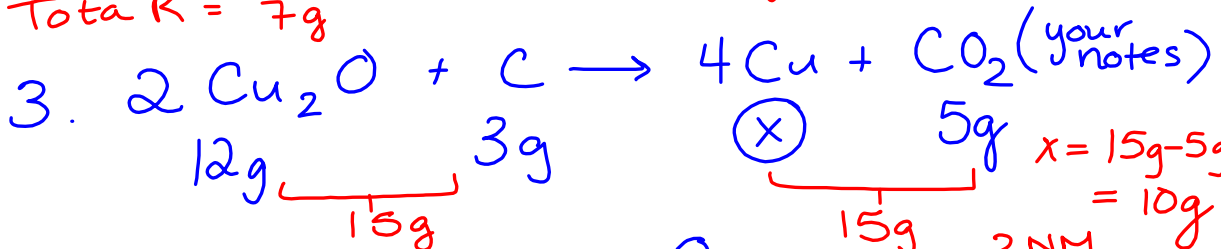
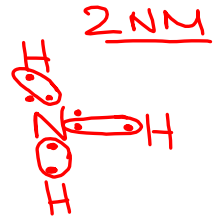
Mass of products = 13g

$13g - 4g = 9g \text{ of } \text{O}_2$

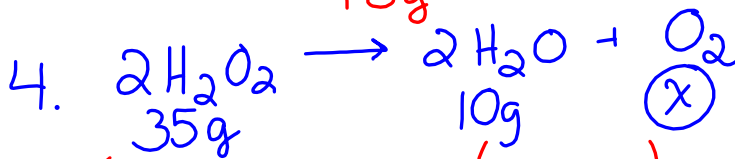


Total R = 7g

x = 7g



x = 15g - 5g = 10g



Total

35g

x = 35g - 10g = 25g

2 NM

